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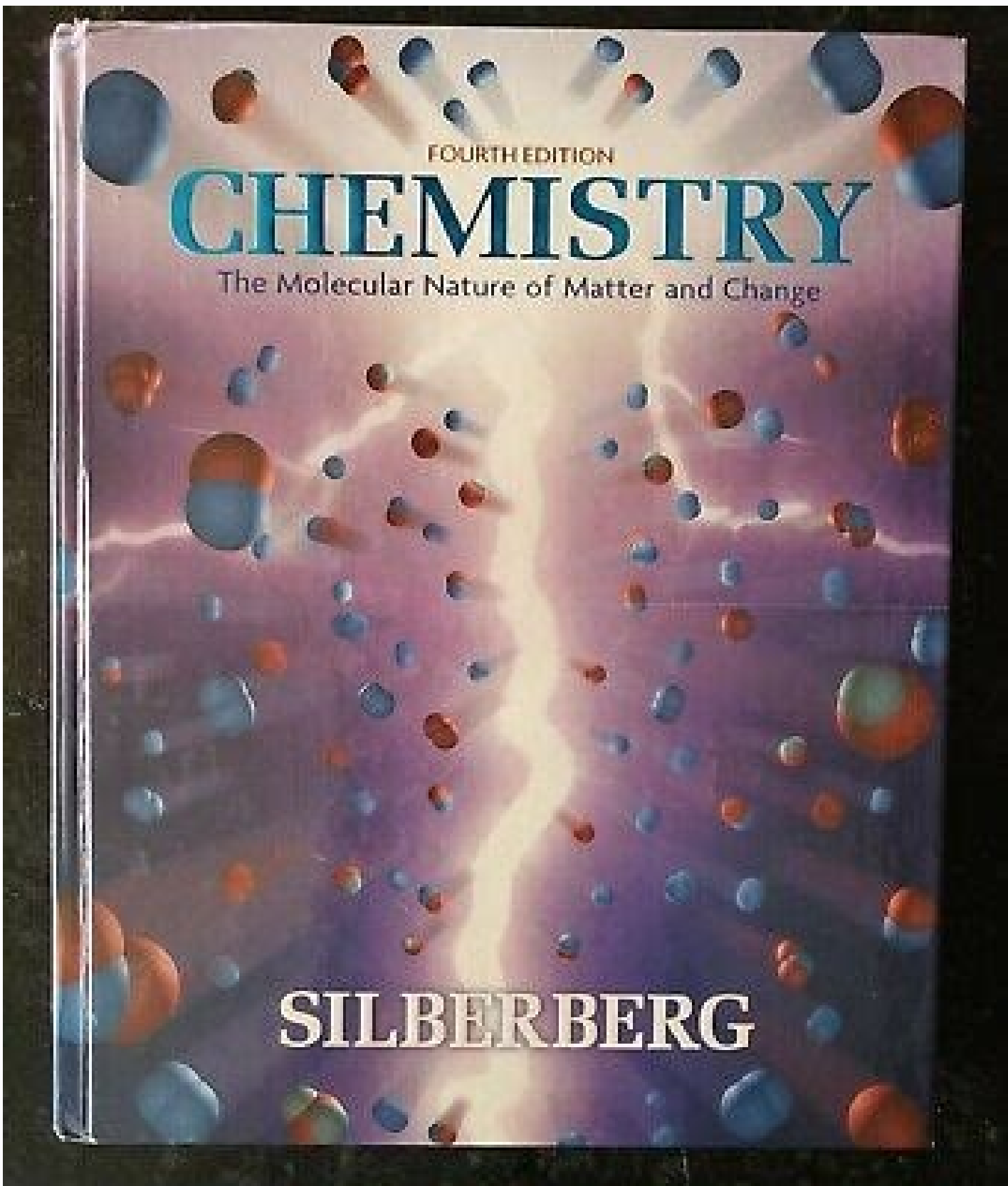
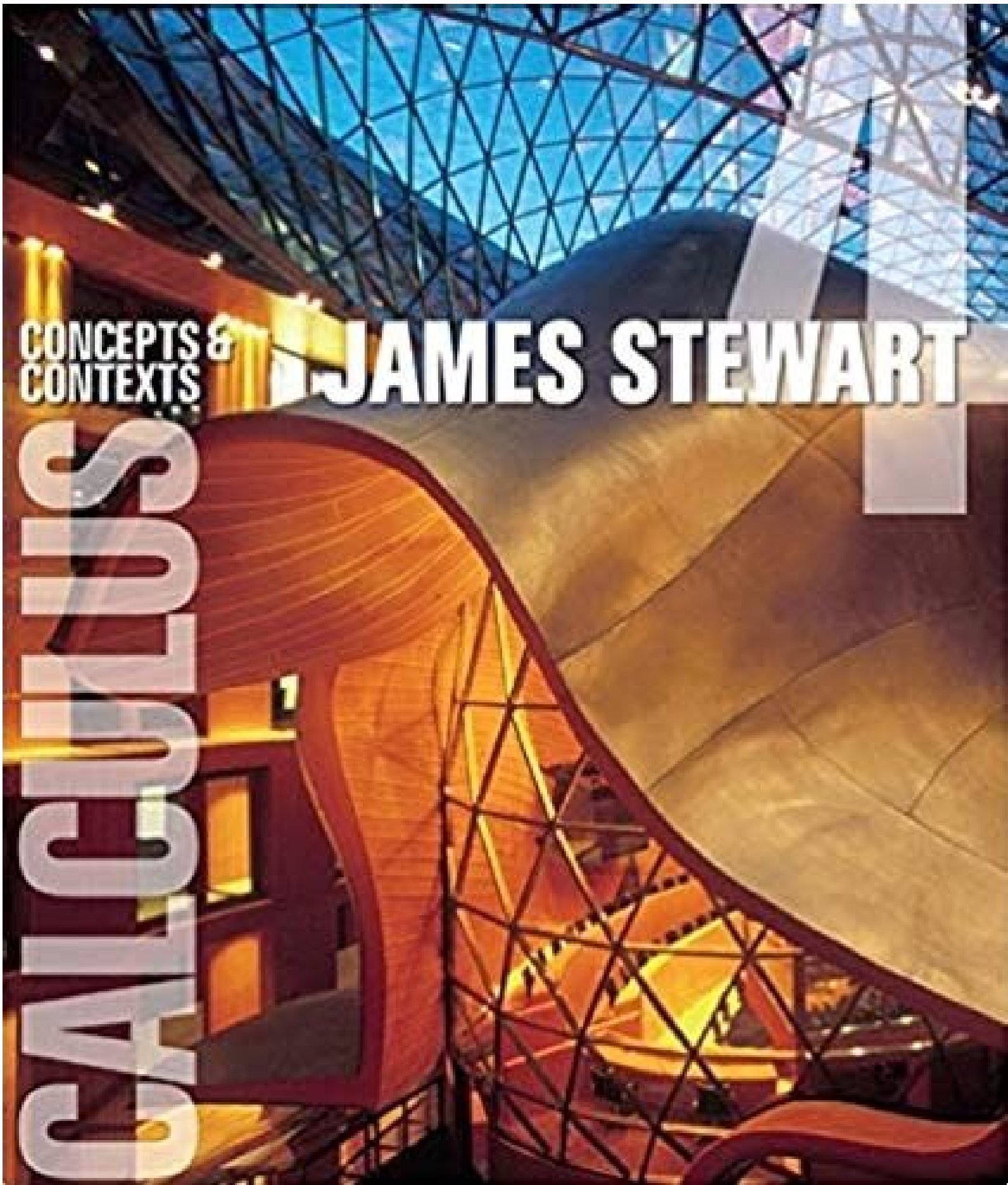
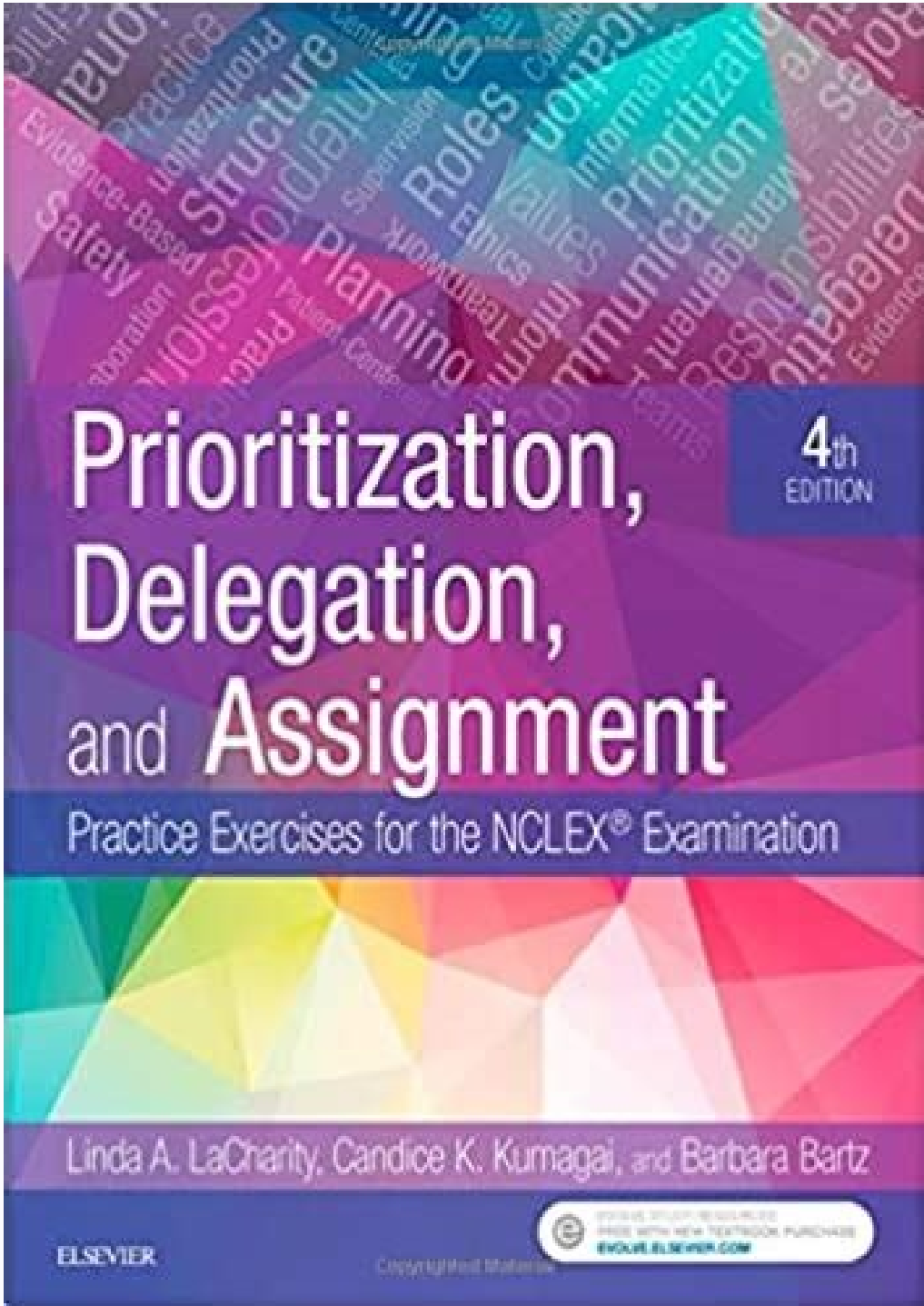
Which of the following correctly illustrates the
Nernst equation for a cell?

☐ (a) $E = E^\circ - \frac{RT}{nF} \ln Q$

☐ (b) $E = E^\circ + \frac{RT}{nF} \ln Q$

☐ (c) $E = E^\circ - \frac{RT}{nF} \ln \frac{1}{Q}$

☐ (d) $E = E^\circ + \frac{RT}{nF} \ln \frac{1}{Q}$



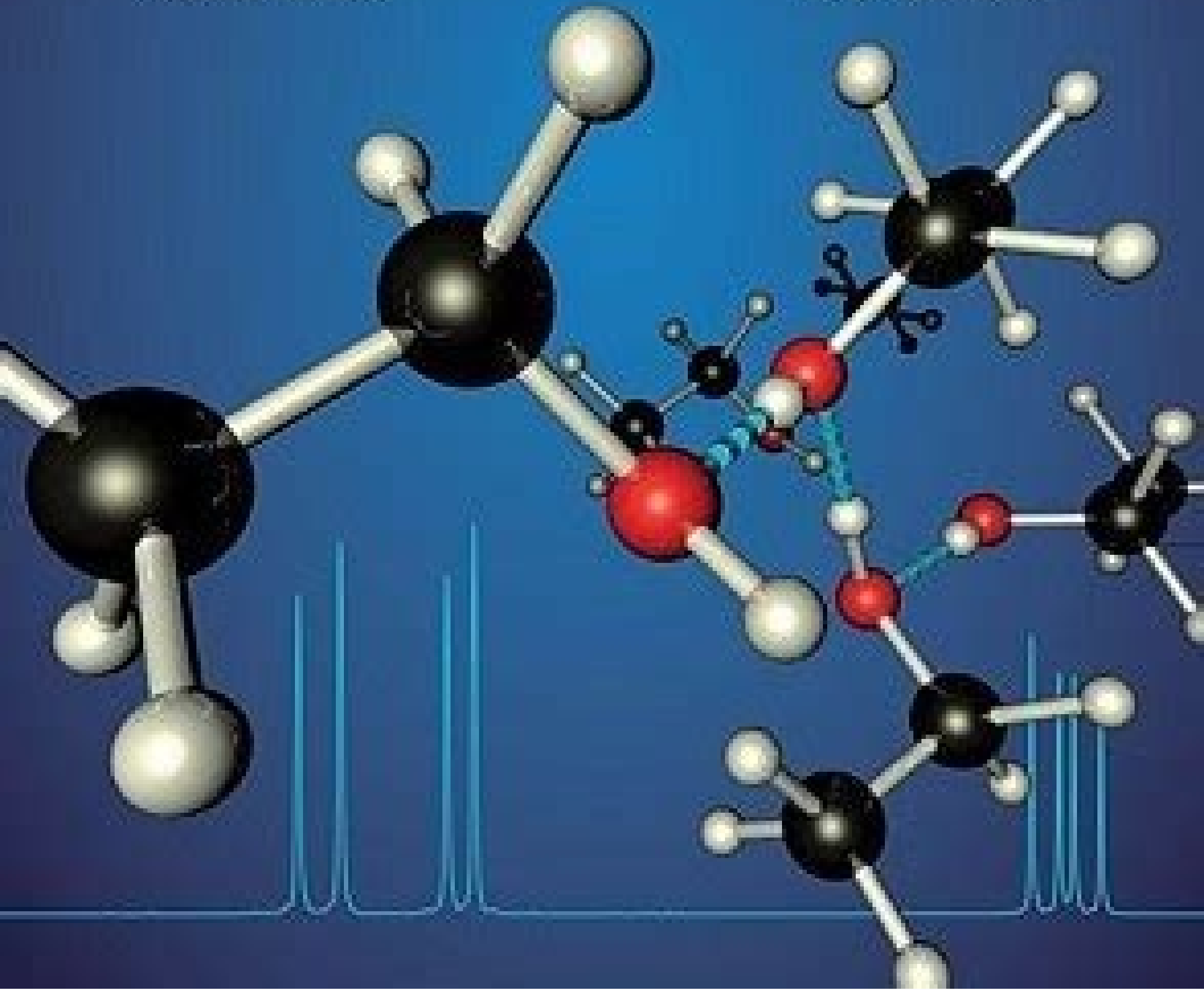
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Chemical Structure and Reactivity

An integrated approach

James Keeler

Peter Wothers



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