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nonmetal    2. Neon                      Nonmetal\_\_\_\_\_ 5. Nitrogen

nonmetal    3. Hydrogen                      Metal\_\_\_\_\_ 6. Calcium

VI. Fill in the blanks using vocabulary terms from unit three.

- Periods \_\_\_\_\_ 1. Horizontal rows on the periodic table are called \_\_\_\_.
- Groups \_\_\_\_\_ 2. Vertical columns on the periodic table are called \_\_\_\_.
- Families \_\_\_\_\_ 3. Groups are often called \_\_\_\_ because the elements behave similarly to each other.
- Metalloid \_\_\_\_\_ 4. If an element has properties of both metals and nonmetals it is known as a \_\_\_\_.
- Ionization Energy \_\_\_\_\_ 5. The amount of energy required to remove an electron from the valence shell of an atom is known as \_\_\_\_.
- Electronegativity \_\_\_\_\_ 6. The tendency for an atom to attract electrons to itself when bonding with another element.
- Ion \_\_\_\_\_ 7. When an atom gains or loses electrons it becomes this.
- Anion \_\_\_\_\_ 8. If an atom gains electrons it becomes a(n) \_\_\_\_.
- Cation \_\_\_\_\_ 9. If an atom loses electrons it becomes a(n) \_\_\_\_.
- Valence Electrons \_\_\_\_\_ 10. Electrons in the outermost energy level are known as \_\_\_\_ electrons.
- Octet Rule \_\_\_\_\_ 11. Atoms react by gaining or losing electrons so as to acquire the stable electron structure of a noble gas. (Usually 8 valence electrons) This is called the \_\_\_\_.
- Atomic Number \_\_\_\_\_ 12. The number of protons in the nucleus of an atom is known as the \_\_\_\_.
- Atomic Mass \_\_\_\_\_ 13. The number of protons and neutrons in the nucleus of an atom is known as the \_\_\_\_.
- D. Mendeleev \_\_\_\_\_ 14. He published the first periodic table.

VII. Trends of the Periodic Table - Write increase or decrease on each line.

| Periodic Table Trend | As you go down the table... | As you go across the table... (left |
|----------------------|-----------------------------|-------------------------------------|
|----------------------|-----------------------------|-------------------------------------|



Public Domain/2012rc/Wikimedia Commons You may be familiar with the peripheric table of the school but therena Ahenck © more than meets the eye with this apparently simple scientific chart. Learn more about the periodic table, including how it was developed and what elements have some interesting history you may not know. There is an Author with a Singular Inspiration you may already know that a man, Dimitri Mendeleev, is © Responsible for creating the periodic table. But what you might not know is © that he was inspired by a deck of cards to write each element on a separate card and organiz e them into groups at the hospital for their atomic weight. CC0/Arcaion/Pixabay Mendeleev was so confident in his system that he was able t o create table space for elements that had not yet been discovered, including giliium. However, your pride too © m had it reject elements initially, as yesterday © lio and other noble gases, which do not fit your structure. CC0/Pexels/Pixabay Items Still Growing Items Still Growing It may seem as though Wej-Weyoun has discovered everything there is to know about the natural world right now, but scientific discoveries happen all the time. New elements are still being added in the healthy © 21st century. Bananas are well known for containing a high concentration of potassium, which is © both an essential nutrient and an element in the periodic table. This fact is © Many times blown out of the proposal, bananas actually contain a slightly radioactive form of potassium. They are completely safe to eat, however, and you should ignore any implications to the contrary. CC0/StockSnap/Pixabay Golden HourCold not © just one element in the periodic table, uh... © also © I'm a color for you. The metal has its yellow hot shade thanks to changing energy levels in elk © Gold thrones, which causes the metal to absorb the blue end of the spectrum of light. 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TRUE - A kg•m 2 /s 2 is a mass unit times a speed squared unit, making it a kinetic energy unit and equivalent to a Joule.. e. FALSE - Work is not dependent ... In a previous unit, it was stated that all objects (regardless of their mass) free fall with the same acceleration - 9.8 m/s/s.This particular aceleration value is so important in physics that it has its own peculiar name - the acceleration of gravity - and its own peculiar symbol - g. But why do all objects free fall at the same rate of acceleration regardless of their mass? Chemistry Scavenger Hunt (pdf)-Internet lesson using the sites listed on the Chemistry page of the Kid Zone. Periodic Tables Online (pdf) - A worksheet I use to review the basics of the periodic table. The sites for this worksheet are listed on the Chemistry links page of the Kid Zone.NOTE: This is the optional page 2 mentioned in the BrainPOP Periodic Table link below. Trends! 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Sep 11, 2021 · Dmitri Medeleev Biography. Dmitri Ivanovich Mendeleev was born on February 8th, 1834, in the Russian province of Siberia. His father was a teacher and his mother opened and operated a glass factory.

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